Cyclisation and Aromatisation of Carotenoids during Sediment Diagenesis

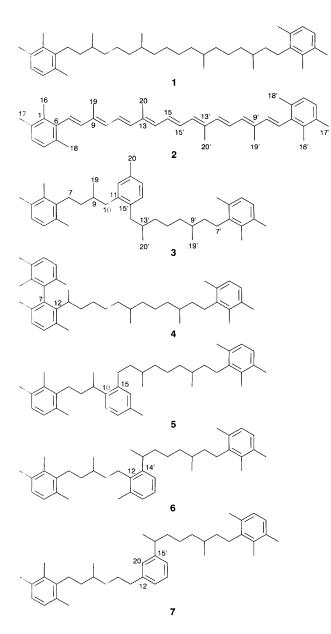
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A novel diaryl isoprenoid with an additional aromatic ring ($C_{40}H_{58}$), formed from the diaromatic carotenoid isorenieratene by cyclisation and aromatisation during sediment diagenesis, is identified in carbonaceous sedimentary rocks.

A number of carbonaceous sediments contain substantial amounts of the diaryl isoprenoid isorenieratane $1,^{1-3}$ which is derived from the diaromatic carotenoid isorenieratene 2, a carotenoid exclusively biosynthesized by photosynthetic green sulfur bacteria (Chlorobiaceae).⁴ Upon GC analyses of the polyaromatic fractions of extracts of a number of sediments (*e.g.* Kimmeridge Clay, Schistes Cartons, Calcaires en Plaquettes, Allgäu Formations) compound **3** elutes just before **1** and possesses a mass spectrum [*m*/*z* 538(20), 173(12), 134(36), 133(100), 119(22%)], suggesting that it is isorenieratane **1** with



an additional aromatic ring formed by cyclisation and aromatisation of the isoprenoidal acyclic part of **1**. The MS, however, does not reveal any strong indications for the position of the additional ring. This is not surprising since MS of 1,2-dialkylbenzenes hardly contain fragment ions due to cleavage of one of the alkyl side chains β to the aromatic ring, but instead are dominated by a fragment formed by loss of both alkyl side chains.⁵ Indeed, the MS of **3** contains a fragment ion at *m*/*z* 119, which is more abundant than in the MS of **1** (22 *vs.* 10%), suggesting the presence of a trialkylbenzene moiety.

The genetic relationship between 1 and 3 was further established by comparison of their ^{13}C content in two sediments (Table 1) as determined by isotope ratio monitoring-GC-MS. Both 1 and 3 are uniquely enriched by *ca*. 14 parts per thousand relative to lipids derived from algae. This is consistent with their derivation from photosynthetic green sulfur bacteria since these organisms fix CO₂ through the reversed TCA cycle leading to biomass anomalously enriched in $^{13}C.^{6}$

To fully determine the structure of 3, it was isolated from the sedimentary rock extract of the Lower Jurassic Allgäu Formation by column chromatography and subsequent reversed phase HPLC using a Polygosil C_{18} preparative column eluting with MeOH-EtOAc (90-10; v/v). The purest fraction (2.3 mg) consisting of 74% of 3 (other 26% consisted of a complex mixture of hydrocarbons, none representing more than 3% of the total fraction) was analysed by ¹H and ¹³C NMR.[†] These data allowed discrimination between the five possible isomers 3-7, which can be formed by cyclisation and aromatisation of 1. From the COSY spectrum of 3 it was deduced that all three aliphatic methyl groups were coupled with protons attached to carbon atoms not α to the aromatic rings as indicated by their chemical shifts in the range δ 1.5–1.8. Furthermore, four CH₂ groups next to an aromatic ring were clearly revealed in the COSY spectrum. Since 3 is the only structure which shows these features, its identity was established. Other shifts are in agreement with this assignment.

The similarity in structure and ${}^{13}C$ content of 1 and 3 establishes a direct link between these sedimentary components and their precursor, isorenieratene 2. The identification of 3 is the first evidence for the process of cyclisation and aromatisation of carotenoids during sediment diagenesis. In fact, the formation of 3 is only one step in this process since components

Table 1 $\delta^{13}C$ (parts per thousand vs. PDB^a) values of selected sedimentary components

Formation name	δ(¹³ C)			
	1	3	Pristane	Phytane
Kimmeridge Clay	-16.6 +0.3 ^b	-16.3+0.2	-30.0 ±0.3	-30.1 ±0.5
Allgäu	-19.0 ±0.4	-19.0 ± 0.3	-33.5 ±0.8	-33.5 ±0.2

^{*a*} PDB indicates PeeDee belemnite, the stable carbon isotope standard. ^{*b*} Standard deviation from three replicates with the isorenieratene carbon skeleton containing up to four additional (condensed) aromatic rings have been identified in sediments.⁷

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Footnote

[†] Selected NMR data for **3**: ¹H (400 MHz, CDCl₃), 2.65 (m, CH₂, C-7'), 2.58 (m, CH₂, C-7), 2.55 (m, CH₂, C-10), 2.41 (m, CH₂, C-14'), 2.23 (s, Me, C-20), 1.86 (m, CH, C-9), 1.68 (m, CH, C-13'), 1.52 (m, CH, C-9'), 1.00 (d, J 6.7 Hz, Me, C-19), 0.97 (d, J 6.7 Hz, Me, C-19'), 0.85 (d, J 6.7 Hz, Me, C-20); Shifts for C-8, C-8', C-10' and C-12' were not determined. ¹³C (100 MHz; CDCl₃) 19.6 (Me, C-19'), 19.5 (Me, C-19), 19.5 (Me, C-20'); these assignments may be interchanged, 21.0 (Me, C-20).

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